Notes

6-DEOXY-8-O-METHYLRABELOMYCIN AND 8-O-METHYLRABELOMYCIN FROM A STREPTOMYCES SPECIES

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In the course of our screening for new antibiotic, new members of benz[a]anthraquinone antibiotics (I and II), which showed antibacterial activity against Gram-positive bacteria were isolated from a culture broth of *Streptomyces tsusimaensis* MI310-38F7. This strain produced nonactin and a minor yellow compound (III) which was identical with X-14881 E^{13} . This report describes the isolation, characterization and structure determination of these compounds, named 6-deoxy-8-O-methylrabelomycin (I) and 8-O-methylrabelomycin (II).

The producing strain MI310-38F7 was cultured at 27°C for 4 days on a rotary shaker (180 rpm) in a 500-ml baffled Erlenmeyer flask containing 110 ml of a production medium consisting of potato starch 3.0%, soybean meal 1.5%, corn steep liquor 0.5%, yeast extract 0.2%, NaCl 0.3%, MgSO₄·7H₂O 0.05%, CaCO₃ 0.3% and CoCl₂·6H₂O 0.001% (pH 7.2 before sterilization). The antibiotic activity in the culture

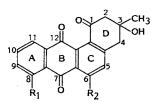
broth was determined by a cylinder plate method using *Micrococcus luteus* IFO 3333 and *Mycobacterium smegmatis* ATCC 607 as the test organism. The isolation procedure is shown in Scheme 2. Further purification was achieved by preparative TLC on silica gel developed with CHCl₃ to give **III** (1.0 mg) and CHCl₃ - MeOH (10:1) to give **I** (13.3 mg) and **II** (14.0 mg).

Physico-chemical properties of I, II and III are summarized in Table 1. The IR spectra of these compounds indicate the presence of ketone carbonyl (I 1703 cm⁻¹, II 1704 cm⁻¹), nonchelated quinone carbonyl (I 1674 cm⁻¹, II 1680 cm⁻¹, III 1660 cm⁻¹) and chelated carbonyl (II 1642 cm⁻¹)²⁰. These spectra are shown in Figs. 1, 2 and 3. These three antibiotics were soluble in CHCl₃, EtOAc and MeOH, but insoluble in water and *n*-hexane.

The ¹H NMR data of I, II and III are shown in Table 2. The ¹³C NMR spectra of I and II are assigned as shown in Table 3.

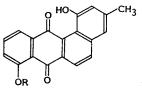
The NMR data indicate that I is the methyl ether of tetrangomycin^{2,3)}. The position of methoxy group was determined from the results of nuclear Overhauser effect (NOE) experiments and the long-range ¹³C-¹H correlation spectroscopy (COSY) spectrum as shown in Fig. 4. Its negative optical rotation was consistent with that of rabelomycin (literature⁴⁾, $-102^{\circ}\pm10^{\circ}$). The absolute configuration at ring D in I was deduced to be the same as rabelomycin. On the basis of these results the structure of antibiotic I was determined to be 6-deoxy-8-*O*-methylrabelomycin.

The molecular formula of II differs from that



6-Deoxy-8-O-methylrabelomycin (I) 8-O-Methylrabelomycin (II) Rabelomycin $\begin{array}{ll} R_1 {=} OCH_3 & R_2 {=} H \\ R_1 {=} OCH_3 & R_2 {=} OH \\ R_1 {=} R_2 {=} OH \end{array}$

Scheme 1.



X-14881 E (III) $R = CH_3$ Tetrangulol R = H

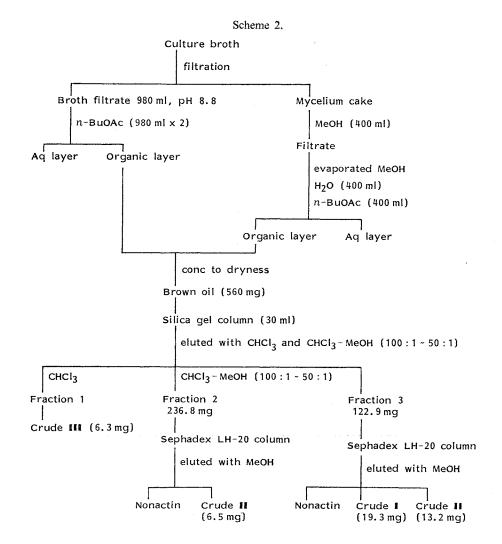
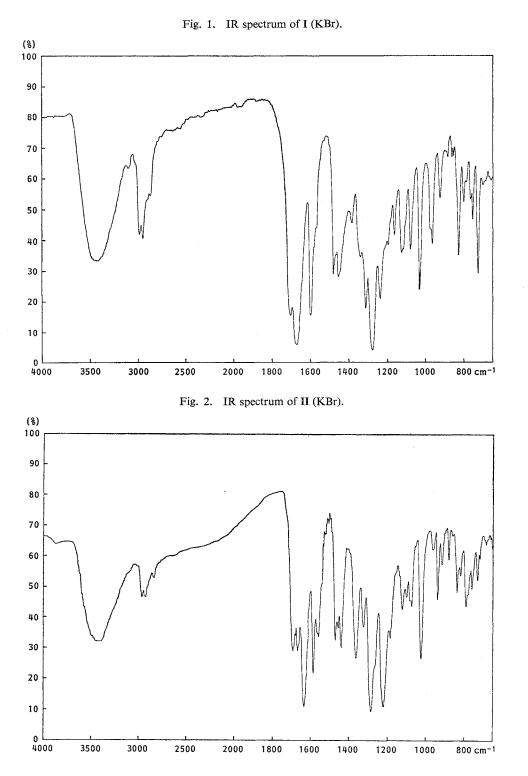


Table 1. Physico-chemical properties of I, II and III.

	I	II	III
Appearance	Yellow needle	Yellow needle	Brown needle
MP (°C)	117~173	191~193	194
Molecular formula	$C_{20}H_{16}O_5$	$\mathbf{C}_{20}\mathbf{H}_{16}\mathbf{O}_{6}$	$C_{20}H_{14}O_4$
HR-MS (m/z) Found:	336.0992	352.0944	318.0865
Calcd:	336.0996	352.0945	318.0880
$[\alpha]_{\rm D}^{28}$	-106.8° (c 0.5, MeOH)	-118.0° (c 0.5, MeOH)	0
UV $\lambda_{\max}^{\text{MeOH}}$ nm (ε)	214 (25,000), 265 (32,900),	224 (23,800), 263 (25,300),	223 (73,000), 308 (35,000),
	375 (5,000)	405 (6,000)	402 (9,000)
$\lambda_{\max}^{MeOH-NaOH}$ nm (c)	214 (36,700), 265 (32,000),	254 (22,800), 316 (8,600),	228 (56,000), 334 (26,000),
	375 (5,000)	465 (4,000)	540 (1,700)
IR $\nu_{\rm max}^{\rm KBr}$ cm ⁻¹	3440, 1703, 1674	3340, 1704, 1680, 1642	3450, 1660
TLC Silica gel (Rf)			
CHCl ₃ - MeOH	0.61	0.68	
(10:1)			
CHCl ₃			0.60

HR-MS: High-resolution mass spectra.



of I by the presence of one additional oxygen atom. The ¹H NMR spectrum of II showed four proton signals in aromatic region, which were

attributable to 5-H, 9-H, 10-H and 11-H. The singlet signal of 5-H (δ 6.92) shifted to higher field than that of I (δ 7.50, d, J=8.0 Hz) and a

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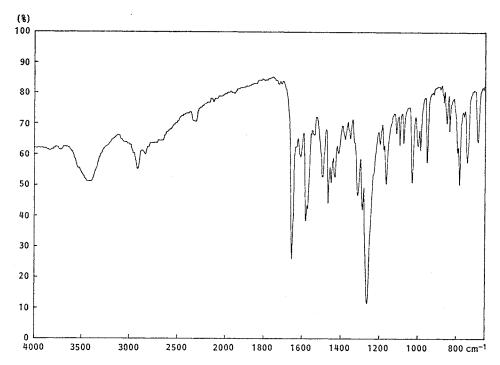


Table 2. ¹H NMR spectrum of I, II and III in CDCl₃ at 400 MHz.

Position	Ι	n	Ш
1-OH			11.15 (s)
$2-H_2$	3.03 (ABq, $J=15.0$ Hz)	2.94 (ABq, J=15.0 Hz)	
2 - H			7.15 (d, $J=1.5$ Hz)
3-CH₃	1.49 (s)	1.45 (s)	2.49 (s)
3-OH	2.29 (s)	2.51 (s)	_
$4-H_2$	3.16 (s)	3.05 (s)	
4 - H	_		7.26 (d, $J=1.5$ Hz)
5-H	7.50 (d, $J = 8.0$ Hz)	6.92 (s)	8.13 (d, J = 8.0 Hz)
6-H	8.26 (d, J = 8.0 Hz)	·	8.31 (d, J = 8.0 Hz)
6-OH		13.03 (s)	
8-OCH₃	4.03 (s)	4.02 (s)	4.08 (s)
9-Н	7.29 (dd, $J=1.5$, 8.0 Hz)	7.30 (dd, $J=1.5$, 8.0 Hz)	7.36 (dd, $J=1.5$, 8.0 Hz)
10-H	7.70 (t, $J = 8.0$ Hz)	7.72 (t, $J = 8.0$ Hz)	7.75 (t, $J = 8.0$ Hz)
11 - H	7.74 (dd, <i>J</i> =1.5, 8.0 Hz)	7.69 (dd, $J=1.5$, 8.0 Hz)	7.95 (dd, $J=1.5$, 8.0 Hz)

signal of hydrogen bonded hydroxyl group appeared at δ 13.03. These results indicate that 6-H in I is replaced by a hydroxyl group in II. NOE experiment indicated vicinal relationship between 9-H (δ 7.30) and 8-OCH₃ (δ 4.02). From the above mentioned results, antibiotic II was determined to be 8-O-methylrabelomycin. II must have the same configuration as rabelomycin since the negative value of optical rotation was almost the same as that of rabelomycin. The NMR spectrum of III indicated that III was related to tetrangulol. The methylene signals at δ 2.94 (2H) and 3.05 (2H) which were observed in the ¹H NMR spectrum of II disappeared in the spectrum of III. The singlet signal of methyl group (δ 2.49) and two new aromatic protons [δ 7.15 (d, J=1.5 Hz), 7.26 (d, J=1.5 Hz)] appeared in the spectrum of III. The presence of the signal at δ 7.26 overlapping with solvent signal was confirmed by spin decoupling

Position	Ι	II
C-1	196.9	195.8
C-2	53.8	53.6
C-3	72.5	71.9
C-4	44.0	44.1
C-4a	146.4	149.5
C-5	133.7	121.8
C-6	130.0	163.8
C-6a	135.1	117.6
	(or 135.2)	
C-7	181.4	188.1
C-7a	120.6	119.6
C-8	159.8	160.2
C-9	117.2	117.4
C-10	135.4	136.3
C-11	119.6	119.8
C-11a	137.7	137.7
C-12	184.6	184.6
C-12a	135.2	137.5
	(or 135.1)	
C-12b	134.2	127.0
3-CH ₃	29.9	. 29.7
8-OCH ₃	56.5	56.5

Table 3. ¹³C NMR chemical shifts of I and II (100 MHz, CDCl₃).

Table 4. The antibacterial spectrum on Mueller-Hinton agar.

Trat anonious	MIC (µg/ml)		
Test organism	I	п	
Staphylococcus aureus Smith	100	12.5	
S. aureus MS9610 (multi-resistant)	100	12.5	
Micrococcus luteus PCI 1001	100	25	
Bacillus subtilis NRRL B-558	100	12.5	
Escherichia coli NIHJ	>100	50	
Shigella dysenteriae JS11910	100	50	
Salmonella typhi T-63	>100	> 50	
Proteus vulgaris OX19	>100	>50	
Serratia marcescens	100	50	
Pseudomonas aeruginosa A3	>100	>25	
Klebsiella pneumoniae PCI 602	>100	> 50	
Mycobacterium smegmatis ATCC 607	>100	>50	
Candida albicans 3147	>100	>50	

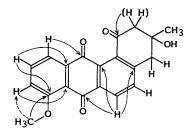
The antibacterial activity of I and II were shown in the Table 4. III was less active than that of I against Gram-positive and Gram-negative bacteria. The acute toxicities (LD_{50} , ip) of I and II in mice were >100 mg/kg and >100 mg/kg, respectively.

We have isolated other *Streptomyces* strains which produced these benz[a]anthraquinone antibiotics: Strain MH897-SF5 produced I and II, strain MI165-34F3 produced only I, strain MI334-21F4 produced rabelomycin.

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Fig. 4. The structure of I.



 \rightarrow : Part of ${}^{3}J_{C-H}$ and ${}^{2}J_{C-H}$ couplings observed in I by the long range ${}^{13}C_{-1}H$ COSY experiment.

 \rightarrow : The NOE arising from irradiation of OCH₃ at δ 4.03.

experiment. A signal of hydrogen bonded hydroxyl group was seen at δ 11.15 which was not observed in the spectrum of I. Therefore, the compound III was found to be X-14881 E¹⁾ on the basis of spectroscopy. Treatment of I with 1 N NaOH gave a dehydrated product, which was identical with III by comparison of their TLC, UV and IR spectra. The presence of III in the culture broth was analyzed by TLC of BuOAc extract which was taken during the fermentation.